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Part 16: 13C-NMR Spectroscopy - Basics and Principle **AQA A-Level Chemistry - Carbon-13 NMR**

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C=O. 150 - 200. The C-13 NMR spectrum for but-3-en-2-one. This is also known as 3-buten-2-one (amongst many other things!) Here is the structure for the compound: You can pick out all the peaks in this compound using the simplified table above. The peak at just under 200 is due to a carbon-oxygen double bond.

interpreting C-13 NMR spectra - chemguide

Peak integration is generally not useful in 13 C-NMR spectroscopy, except when investigating molecules that have been enriched with 13 C isotope (see section 5.6B). The resonance frequencies of 13 C nuclei are lower than those of protons in the same applied field - in a 7.05 Tesla instrument, protons resonate at about 300 MHz, while carbons resonate at about 75 MHz.

C-13 NMR - Chemistry LibreTexts

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We have seen that 13 C NMR is usually decoupled and therefore there is no splitting of signals which limits the information we can get as to how many hydrogens are connected to a carbon atom. And even combining 1 H and 13 C NMR spectra may not give a definite evidence for choosing only one structure.

DEPT NMR: Signals and Problem Solving - Chemistry Steps

Next functional group recognizable in 1 H NMR spectroscopy is the ~10 ppm signal of aldehydes. Usually, it shows up as a singlet, though splitting with adjacent protons is not uncommon either: Another way of identifying aldehydes and ketones as well as checking for a ~200 ppm signal on the 13 C NMR. Esters and carboxylic acids appear less downfield - 160-180 ppm.

NMR Spectroscopy Practice Problems - Chemistry Steps

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Spectroscopy Problems. In each of these problems you are given the IR, NMR, and molecular formula. Using this information, your task is to determine the structure of the compound. The best approach for spectroscopy problems is the following steps: Calculate the degree of unsaturation to limit the number of possible structures.

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