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molecule = $23\ 842\ 610 = 140.33\ 10\ 23$
 $3 = 1.4\ 10\ 21\ \text{g}$ 5. Atomic mass = $10\ 19$
 $81\ 11\ 100 = 1081\ 100 = 10.81\ 6$.
Average atomic mass = $200\ 90\ 199\ 8$
 $202\ 2\ 100 = 199.96\ 200\ 7$. Molecular
mass of $\text{NH}_3 = 14 + (3\ 1) = 17$ Number
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Total Marks : 200 {Chemistry 50 marks,

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Duration: 90 minutes for each paper:
Type of questions: Multiple Choice Questions (MCQs) Total Marks: 200:
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