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3.73×10^{-3} The data in Table $\backslash(\backslash\text{PageIndex}\{1\}\backslash)$ were obtained by removing samples of the reaction mixture at the indicated times and analyzing them for the concentrations of the reactant (aspirin) and one of the products (salicylic acid).

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Section Quiz: Scientific Method In the space provided, write the letter of the term or phrase that best completes each sentence or best answers each question. 1. Which of these observations is qualitative rather than quantitative? a. A chemical reaction is complete in 2.3 s. b. The solid has a mass of 23.4 g. c. The density of aluminum is 2.70 g/cm³ ...

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Modern Chemistry 107 Chapter Test Name Class Date Chapter Test A, continued ____18. Which type of mixture contains the smallest particles?

a. emulsions b. solutions c. suspensions d. colloids ____19. ... 15. a16. 17. b 18. b) Modern Chemistry ...

Assessment Chapter Test A

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

D. are the farthest to the right in the periodic table. 3. The periodic law states that A. the chemical properties of elements can be grouped according to periodicity. B. the properties of the elements are functions of atomic mass. C. all elements in the same group have the same number of valence electrons.