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temperature and pressure at which one mole of gas occupies a volume of 22.4 L. equal volumes of gases at the same temperature and pressure contain equal numbers of particles.

### Quia - Chapter 10 "Chemical Quantities" Vocab

Chapter 10 Chemical Quantities91 SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287–296) This section defines the mole and explains how the mole is used to measure matter. It also teaches you how to calculate the mass of a mole of any substance.

### SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287-296)

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