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Nuclear Equations Answer Key - bitofnews.com

c- -1. beta minus. Bombarding plutonium-239 with an alpha particle creates an isotope of curium. Balance the equation. 239/94Pu + 4/2He → (a)/(b)Cm + (c)/(d)n. Identify the values for the unknown letters. a - 242. b - 96. c - 1.

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Balancing Nuclear Equations: Matter must be conserved including all p+ & n°. Example: Decay reaction (α decay) 215! 86 219Rn" 2 4He+ 84 Po Fission Reaction ! 0 1n+ 92 235U" 36 92Kr+ 56 141Ba+3 0 1n Fusion Reaction: ! 17 35Cl+ 1 1H" 18 36Ar another example! 1 2H+ 1 3H" 2 4He+no Practice Fill in the missing symbol and name the reaction: 1. ! 1 3H" ____ + #1 0e 2. ! 92

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Nuclear Equations Worksheets - Teacher Worksheets

Radiation Science and Engineering Center The Pennsylvania State University University Park, PA 16802 1. Balancing Nuclear Equations. When balancing nuclear equations, the sums of the atomic and mass numbers must be the same on both sides of the equation. In some cases one of the symbols in the list below will be used to complete the equation.

Balancing Nuclear Equations - Texas A&M University

A balanced nuclear reaction equation indicates that there is a rearrangement during a nuclear reaction, but of subatomic particles rather than atoms. Nuclear reactions also follow conservation laws, and they are balanced in two ways: The sum of the mass numbers of the reactants equals the sum of the mass numbers of the products.

21.2 Nuclear Equations - Chemistry

The first step to balance the equation is to write down the chemical formula of reactants that are listed on the left side of the chemical equation. After this, you can list down the products on the right hand side of the chemical equation. There is an arrow between the sides, signaling the direction the

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