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Titration. Measure the quantity of a known solution needed to neutralize an acid or base of unknown concentration. Use this information to calculate the unknown concentration. A variety of indicators can be used to show the pH of the solution. 5 Minute Preview.

Titration Gizmo : Lesson Info :

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School. Name: _ Date:
_ Student Exploration:
Titration Vocabulary:
acid, analyte, base,

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Measure: A titration can be used to determine the concentration of an

acid or base by measuring the amount of a solution with a known concentration, called the titrant, which reacts completely with a solution of unknown concentration, called the analyte. The point at which this occurs is called the equivalence point.

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 concentration of an
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 known concentration,
 called the titrant,
 which reacts
 completely with a
 solution of unknown
 concentration, called
 the analyte.

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Titration Answer Key

Answers To Student
 Exploration Titration
 Gizmo Gizmo Warm-up
 Litmus is an example
 of an indicator, a
 substance that
 changes color
 depending on its pH
 (pH is a measure of the
 concentration of...

Titration Gizmo Answer Key - The Forward

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will use indicators to
show how acids are
neutralized by bases,
and vice versa. To
begin, check that 1.00
M NaOH is selected for
the Burette,... (Activity
C continued on next
page) Activity C
(continued from
previous page) 3.

**Titration Gizmo
Answer Key Activity
C Continued**

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