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Notes Chapter 5 Surface. Adsorption: (i) The accumulation of molecular species at the surface rather than in the bulk of a solid or liquid is termed as adsorption. (ii) It is a surface phenomenon. (iii) The concentration of adsorbate increases only at the surface of the adsorbent.

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Chemistry Notes for class 12 Chapter 5 Surface Chemistry Surface Chemistry is the branch of chemistry which deals with the phenomenon that occurs on the surfaces or interfaces, such phenomenon

includes corrosion, catalysis, crystallisation, etc Adsorption Due to unbalanced attraction forces, accumulation of molecular species at the surface rather

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Class 12 Chemistry Surface Chemistry. Introduction. Introduction. Surface Chemistry deals with the study of physical and chemical phenomena occurring at the boundary (interface) separating two bulk phases. The bulk phase can be a pure compound or a solution.

CBSE NCERT Notes Class 12 Chemistry Surface Chemistry

Surface Chemistry Class 12 Notes

Chemistry Chapter 5. 1. This phenomenon of attracting and retaining the molecules of a substance by a solid (or a liquid) on its surface resulting into a higher concentration of the molecules on the surface is known as adsorption. 2.

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phenomena that occur at the interface of
two phases, including solid-liquid
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solid-vacuum interfaces, and liquid-gas
interfaces. It includes the fields of
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