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12. Inert gasses are arranged in what part of the Periodic Table? Far right group or column. PART III - Chemical Reactions, Reactants and Products. 13. A liquid that forms a gas when heated is an example of a chemical reaction. False 14. When Vinegar (acetic acid CH3COOH) reacts with baking

soda (NaHCO3), one of the products is Carbon Dioxide (CO2). True 15.

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molecule on the reactant side. Because we added HCl and NaOH in steps 1-4, the colours changed back to their original shades to maintain equilibrium. 7) BaCr 2 O 7 is much more soluble than BaCrO 4, as the K 2 Cr 2 O 7 solution was darker in colour when mixed with Ba(NO 3) 2. 8) No precipitate formed in step 7 of Part IV because H in HCl was added, shifting the + reaction to the reactant side.

Experiment with Equilibrium - Lab 12 A Investigating ...

Instructional design and development of written and interactive content: Kognito Interactive (www.kognito.com) Main content contributor and reviewer:

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