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2 Measurements and Calculations

CHAPTER 2 REVIEW Measurements and Calculations SECTION 2-3 SHORT ANSWER Answer the following questions in the space provided. 1. Report the number of significant figures in each of the following values: a. 0.002 37 g d. 64 mL b. 0.002 037 g e. 1.3 102 cm c. 350. J f. 1.30 102 cm 2. Write the value of the following operations using scientific notation: a. 103 1 0 1 2 0 6 b. 8 2 1 1 0

2 Measurements and Calculations

2(g) → NO 2(g) a. Balance the equation. 2N 2O(g) 3O 2(g) → 4NO 2(g) 4 mol NO 2:3 mol O 2 b. What is the mole ratio of NO 2 to O 2? 15.0 mol c. If 20.0 mol of NO 2 form, how many moles of O 2 must have been consumed? True d. Twice as many moles of NO 2 form as moles of N 2O are consumed. True or False? False e. Twice as many grams of NO 2 ...

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PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 8. 9.7 10¹⁴ Hz What is the frequency

of light that has a wavelength of 310 nm? 9. 9.4 910 m What is the wavelength of electromagnetic radiation if its frequency is 3.2 10² Hz? Wavelength Wavelength Wave A Wave B 26 ARRANGEMENT OF ELECTRONS IN ATOMS ...

4 Arrangement of Electrons in Atoms

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b ...

6 Chemical Bonding

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8.

_____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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