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One mole of iron has a mass of 56 g.  
 Moles of compounds A mole of a molecular compound contains  $6 \times 10^{23}$  molecules. It has a mass that is equal to its relative formula mass.

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 Directions: Determine the number of moles in each of the quantities below.  
 1) 52 g of NaCl. ANSWER: 0.89 mole  
 Chemistry IF8766. Page 50.

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Reveal answerupdown.  $M_r$  of NaOH =  $23 + 16 + 1 = 40$ .  $M_r$  of  $\text{Na}_2\text{SO}_4 = 23$

+ 23 + 32 + 16 + 16 + 16 + 16 = 142.  
 Number of moles of NaOH = mass ÷  
 relative formula mass = 20 ÷ 40 = 0.5  
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 Directions: Determine the number of  
 moles in each of the quantities below. 1)  
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 ... mole = molecular weight / mass  
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 / mole. As  $1.626 \times 10^{23}$  molecules of  
 NaOH is also equal to 0.27 moles, and

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