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Section 8.2 - The Nature of Covalent Bonding In ionic bonding, atoms transfer electrons to achieve noble gas configuration. In covalent bonding, atoms share electrons to achieve noble gas configuration. Most atoms share electrons until they have a total of 8 valence

8.2 the nature of covalent bonding

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Terms in this set (21) When atoms share
electrons to gain the ____ configuration
of a noble gas, the bonds formed are
____. (covalent, unshared pairs, shared,
stable electron, single) Stable ...

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8.2 The Nature of Covalent Bonding >
8.2 The Nature of Covalent Bonding > 8
Hydrogen gas consists of diatomic
molecules whose atoms share only one
pair of electrons, forming a single
covalent bond. The Octet Rule in The
Octet Rule in Covalent Bonding Covalent
Bonding Single Covalent Bonds

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Covalent Bonding ...**

8.2 The Nature of Covalent Bonding -
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conduct chemistry. That's because
unlike humans, the rest of nature
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water as a solvent, uses a subset of the
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8.1 Molecular Compounds 8.2 The

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